

# Chemical Speciation of PM<sub>2.5</sub> in the Treasure Valley, Idaho, USA

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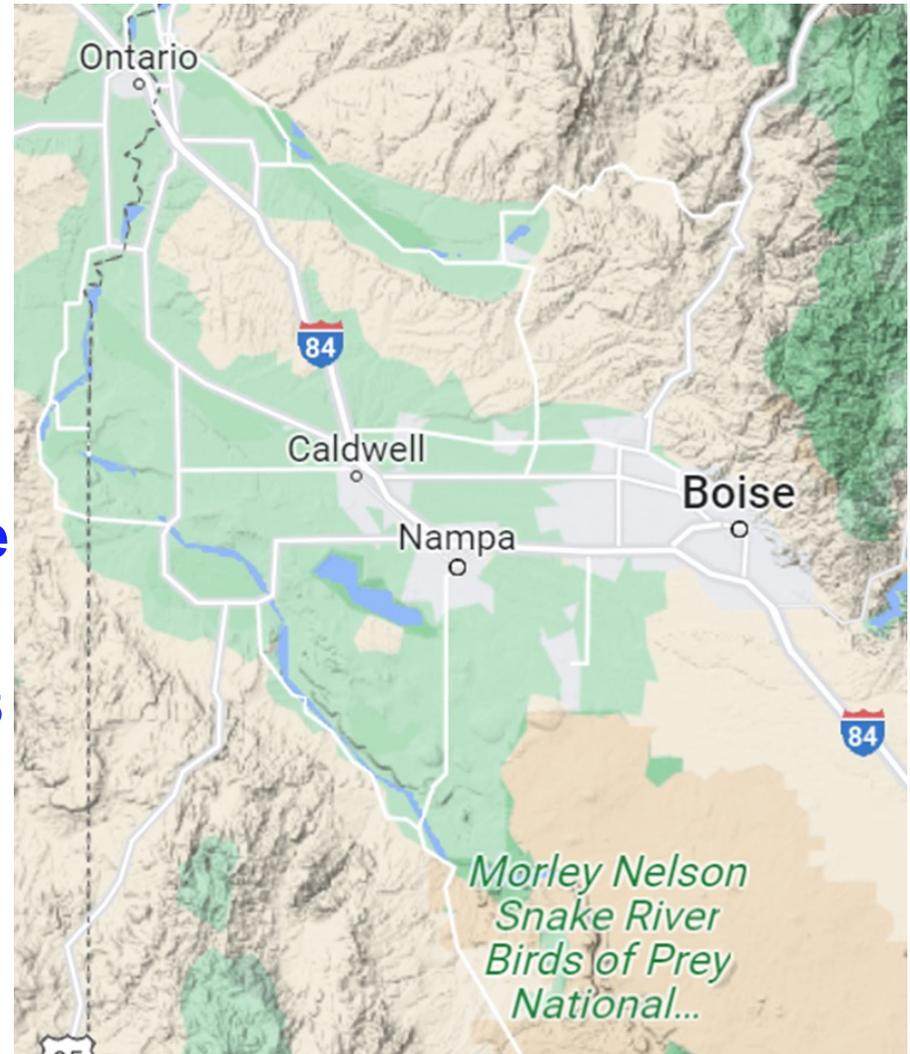


# Background

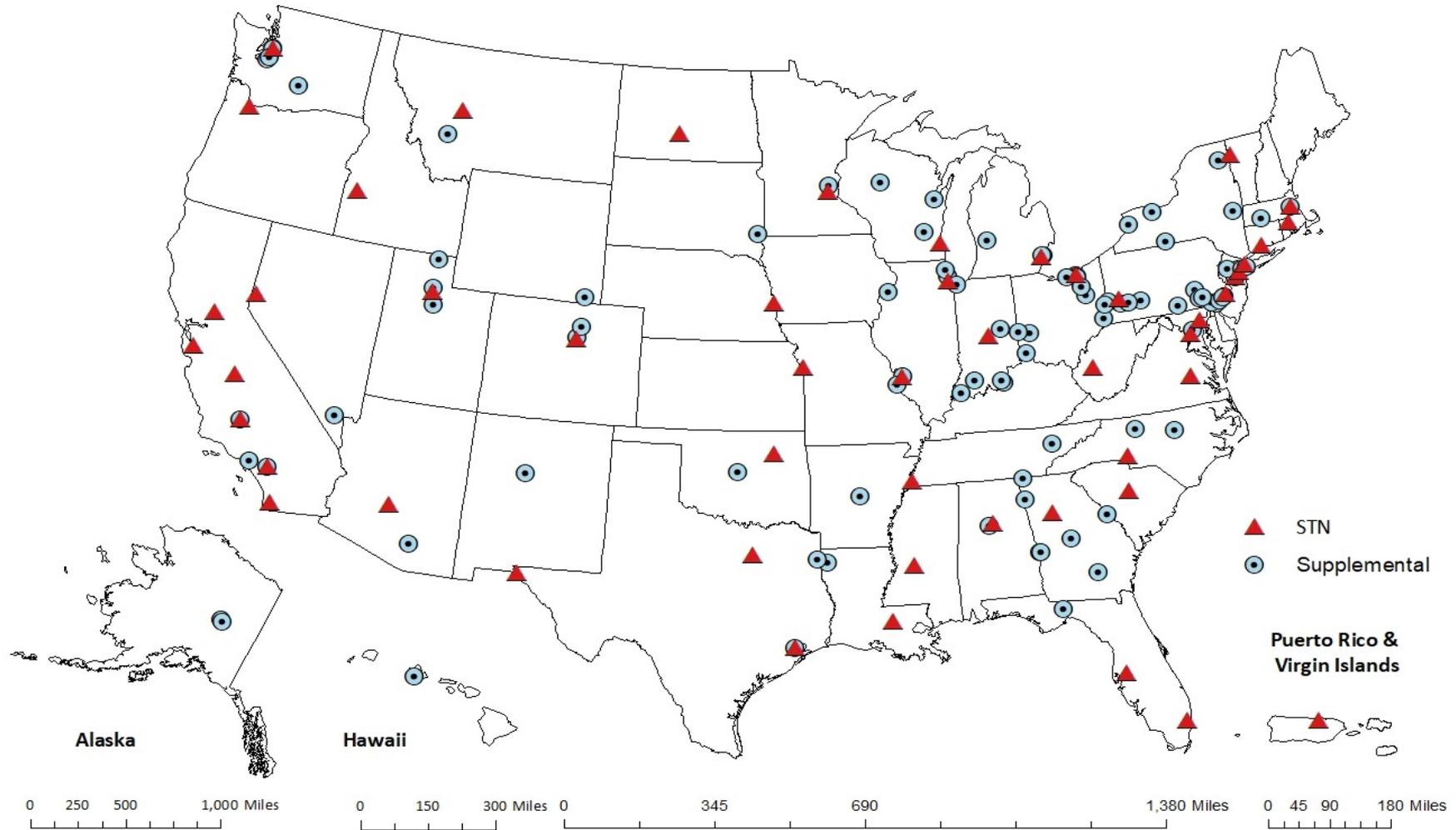
- **PM<sub>2.5</sub> and Ozone: Adverse health effects**
- **EPA: National Ambient Air Quality Standards (NAAQS), The Clean Air Act**
- **EPA is considering to re-set tighter air quality standards (NAAQS)**
- **More nonattainment areas**

# Plans for the Treasure Valley

- The most populous metropolitan area in Idaho (a combined population of ~750k)
- SIP: Removal of Inspection and Maintenance program
- Sources of  $PM_{2.5}$  and ozone
- Speciation of  $PM_{2.5}$
- Receptor modeling of  $PM_{2.5}$
- Regional Chemical Transport Modeling: ozone and  $PM_{2.5}$
- Control measures



# Chemical Speciation Network (CSN)



- ~50 STN sites: use the national contract for shipping, handling and analysis of samples for consistency
- ~100 supplemental sites: have the option of using other state, local or contract laboratories for analysis



Ontario

84

Caldwell

Boise

Nampa

St Luke's

84

*Morley Nelson  
Snake River  
Birds of Prey  
National...*

# Chemical Analyses and Calculations

- X-ray fluorescence; PTFE filter: elements (e.g. Al, Si, Ca, Fe, Ti, Ni, Cu, Zn, As, Mg, Sc, Na, Cl, V, Cr, Se, Br, Rb, Sr, Yt, Cs, Ba, La, Ce, Sm, Eu, Tb, Pb, K, ....)
- Ion chromatography; nylon filter: e.g.  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{NH}_4^+$ , and  $\text{Na}^+$ ,  $\text{Cl}^-$
- Thermal optical analysis (TOA); quartz filter: Carbon species (e.g. OC1, OC2, OC3, OC4, OP, EC1, EC2, EC3)

# Chemical Analyses and Calculations

$$\text{Soil} = 2.2 * \text{Al} + 2.49 * \text{Si} + 1.63 * \text{Ca} + 2.42 * \text{Fe} + 1.94 * \text{Ti}$$

$$\text{Salt} = 1.8 * \text{Cl}$$

$$\text{AS} = 4.125 * \text{S}$$

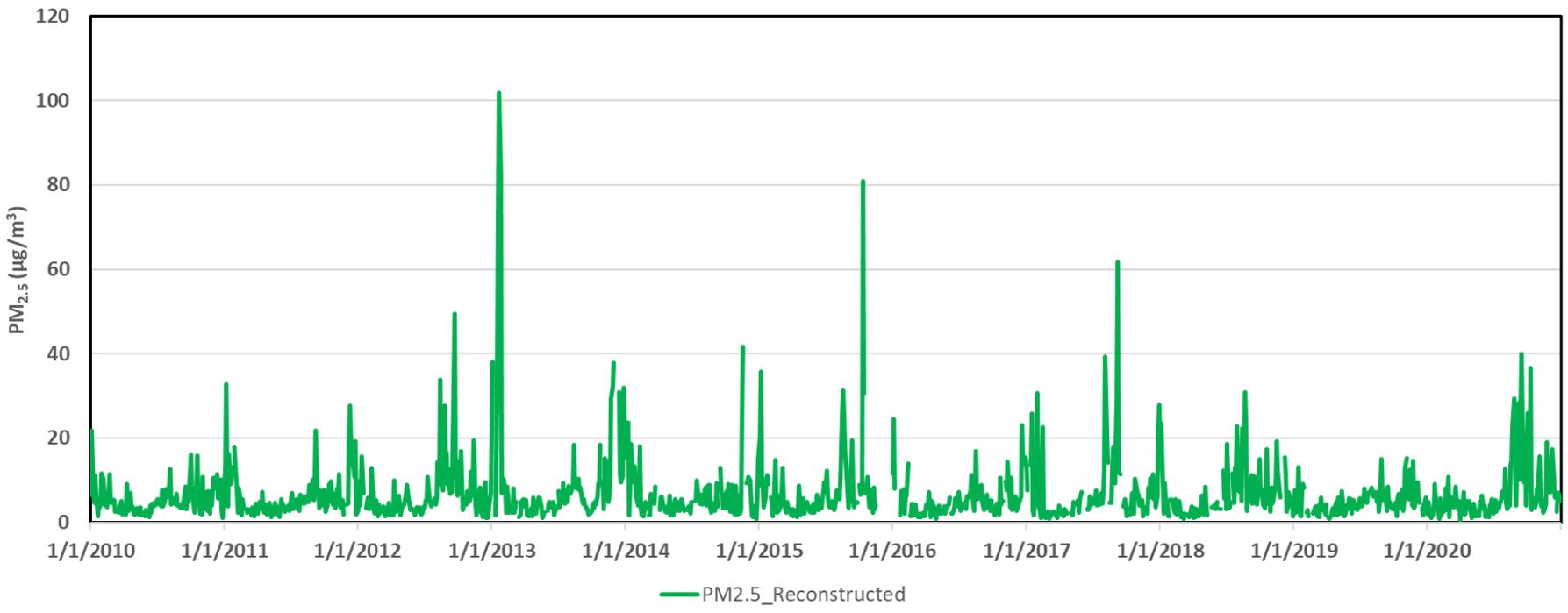
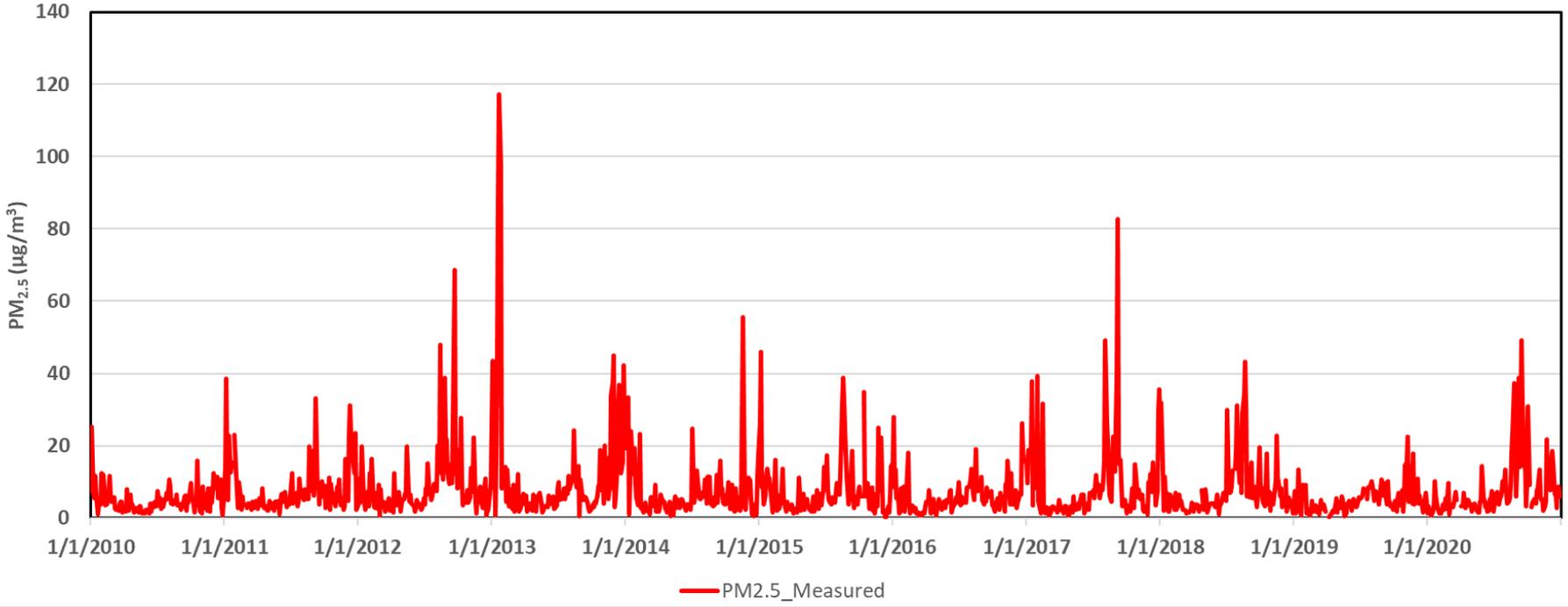
$$\text{AN} = 1.29 * \text{NO}_3$$

$$\text{OC} = \text{OC1} + \text{OC2} + \text{OC3} + \text{OC4} + \text{OP}$$

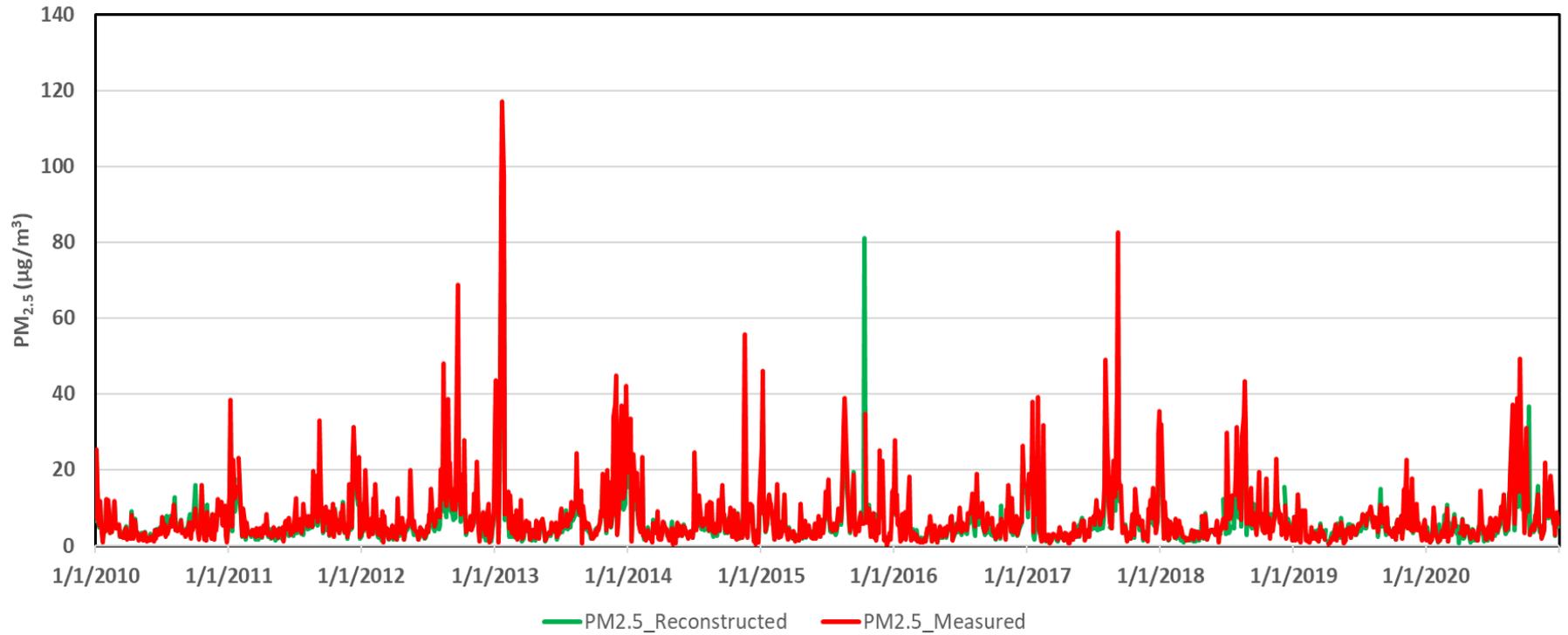
$$\text{EC} = \text{EC1} + \text{EC2} + \text{EC3} - \text{OP}$$

$$\text{OM} = 1.4 * \text{OC}$$

$$\text{Reconstructed PM}_{2.5} = \text{OM} + \text{EC} + \text{Soil} + \text{Salt} + \text{AS} + \text{AN}$$



-13.53%



# Limitations & artifacts

- Ammonium sulfate is calculated with an assumption that all sulfur is present as ammonium sulfate
- 100% neutralization
- Particle-bound water is not measured
- Many species were not included in the calculations

# Calculations

$$\text{DON} = (\text{NH}_{4,\text{SO}_4}) / \text{SO}_4,$$

$$\text{NH}_{4,\text{SO}_4} = \text{NH}_4 - 0.29 * \text{NO}_3$$

$$\text{AS} = (\text{DON} + 1) * \text{SO}_4$$

The ionic fractions:

$$S = \text{SO}_4 / (\text{SO}_4 + \text{NO}_3 + \text{NH}_4);$$

$$N = \text{NO}_3 / (\text{SO}_4 + \text{NO}_3 + \text{NH}_4);$$

$$A = \text{NH}_4 / (\text{SO}_4 + \text{NO}_3 + \text{NH}_4);$$

## Particle-Bound Water (PBW)

(MATS, 2014); based on results of Aerosol Inorganic Model (AIM)

DON  $\leq$  0.225

$$\begin{aligned} \text{PBW} = \{ & 595.556 - 1440.584 * S - 1126.488 * N + 283.907 * (S^{**1.5}) - 13.384 * (N^{**1.5}) \\ & - 1486.711 * (A^{**1.5}) + 764.229 * (S^{**2}) + 1501.999 * (N * S) \\ & + 451.873 * (N^{**2}) - 185.183 * (S^{**2.5}) - 375.984 * (S^{**1.5}) * N \\ & - 16.895 * (S^{**3}) - 65.814 * (N^{**1.5}) * S + 96.825 * (N^{**2.5}) \\ & + 83.037 * (N^{**1.5}) * (S^{**1.5}) - 4.419 * (N^{**3}) + 1720.818 * (A^{**1.5}) * S \\ & + 1220.383 * (A^{**1.5}) * N - 311.496 * (A^{**1.5}) * (S^{**1.5}) \\ & + 148.771 * (A^{**1.5}) * (N^{**1.5}) + 1151.648 * (A^{**3}) \} * (\text{SO}_4 + \text{NO}_3 + \text{NH}_4); \end{aligned}$$

DON  $>$  0.225:

$$\begin{aligned} \text{PBW} = \{ & 202048.975 - 391494.647 * S - 390912.147 * N + 442.435 * (S^{**1.5}) \\ & - 155.335 * (N^{**1.5}) - 293406.827 * (A^{**1.5}) + 189277.519 * (S^{**2}) \\ & + 377992.610 * N * S + 188636.790 * (N^{**2}) - 447.123 * (S^{**2.5}) \\ & - 507.157 * (S^{**1.5}) * N - 12.794 * (S^{**3}) + 146.221 * (N^{**1.5}) * S \\ & + 217.197 * (N^{**2.5}) + 29.981 * (N^{**1.5}) * (S^{**1.5}) - 18.649 * (N^{**3}) \\ & + 216266.951 * (A^{**1.5}) * S + 215419.876 * (A^{**1.5}) * N \\ & - 621.843 * (A^{**1.5}) * (S^{**1.5}) + 239.132 * (A^{**1.5}) * (N^{**1.5}) \\ & + 95413.122 * (A^{**3}) \} * (\text{SO}_4 + \text{NO}_3 + \text{NH}_4) \end{aligned}$$

# Reconstructed PM<sub>2.5</sub>

$$\text{PM}_{2.5} = \text{OM} + \text{EC} + \text{Soil} + \text{Salt} + \text{AN} + \text{AS} + \text{PBW} + \text{NSS} + \text{OPP}$$

Where

PM<sub>2.5</sub> = PM<sub>2.5</sub> mass

OM = organic mass

EC = elemental carbon

Soil = oxides of crustal elements

Salt = road salt

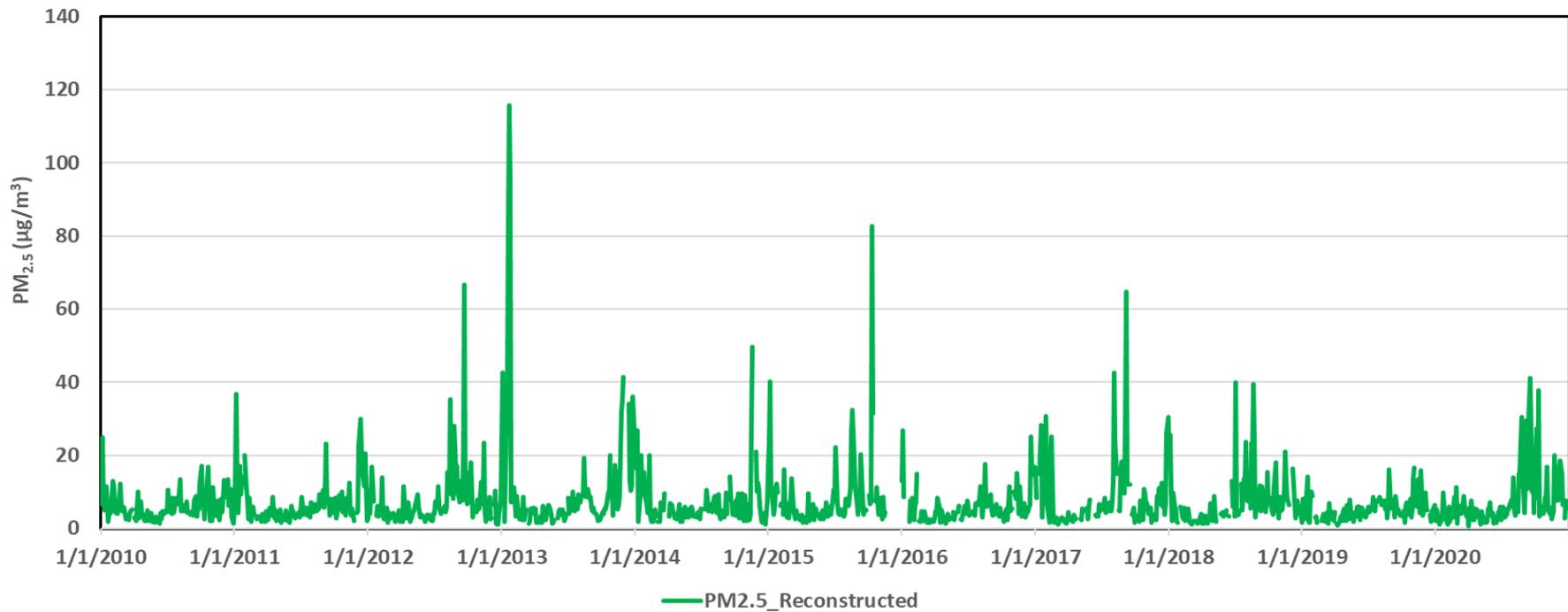
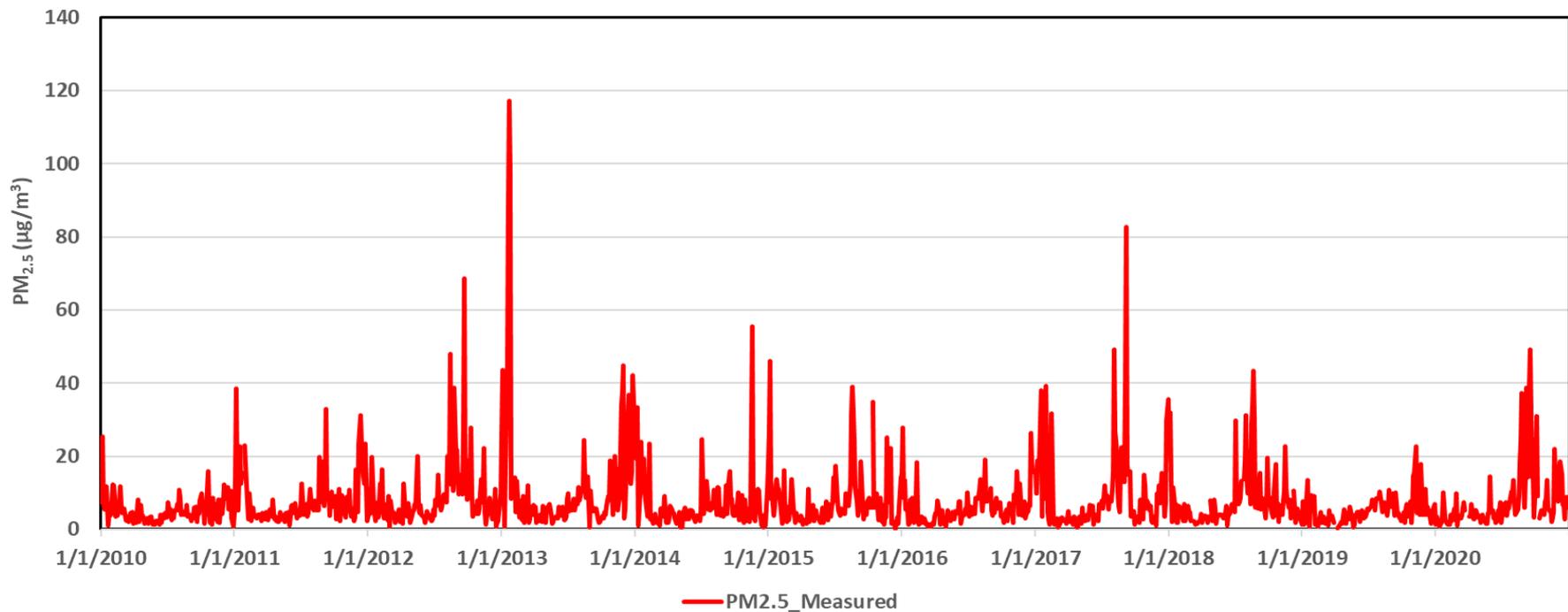
AN = ammonium nitrate

AS = ammonium sulfate

PBW = particle-bound water

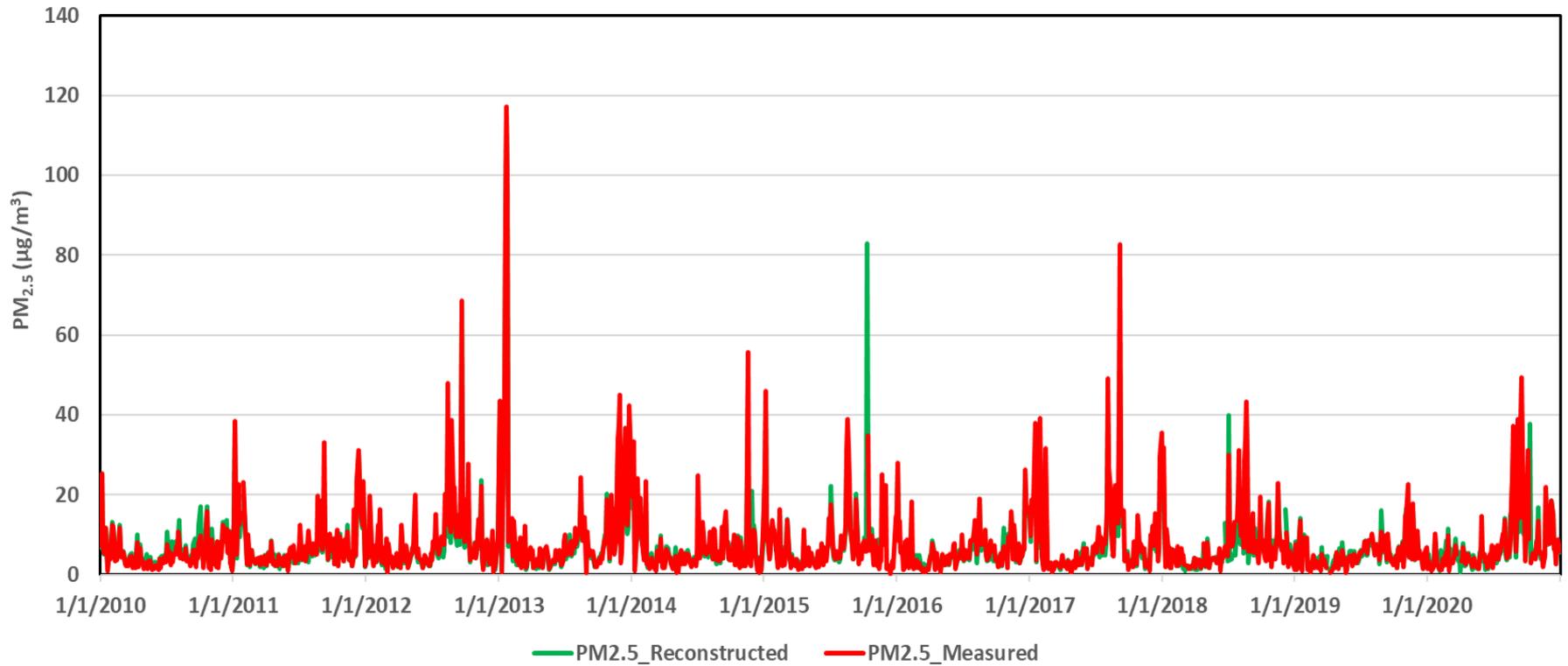
NSS = non-sulfate sulfur                       $\text{NSS} = \text{S} - 0.338 * \text{SO}_4$

OPP = other primary particulate {Mg, Ni, Cu, Zn, As, Pb ...}

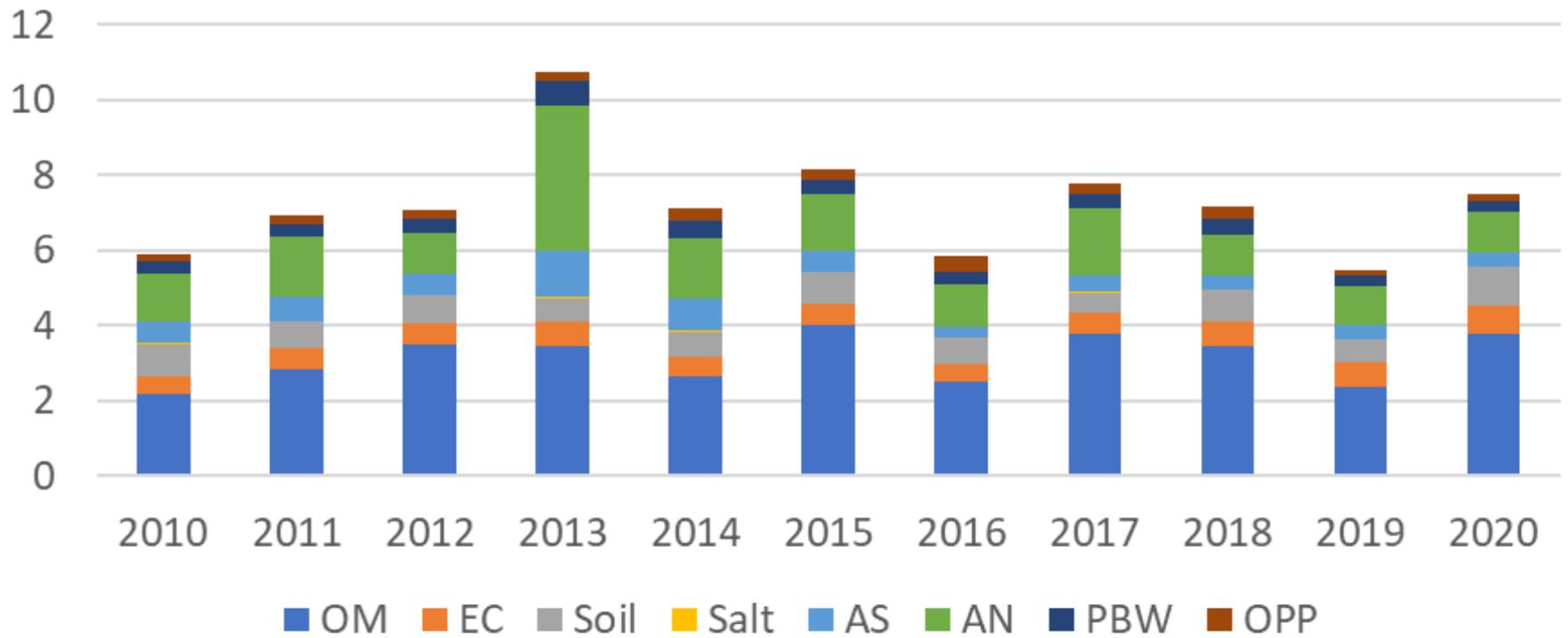


# Comparison of reconstructed and measured PM<sub>2.5</sub>

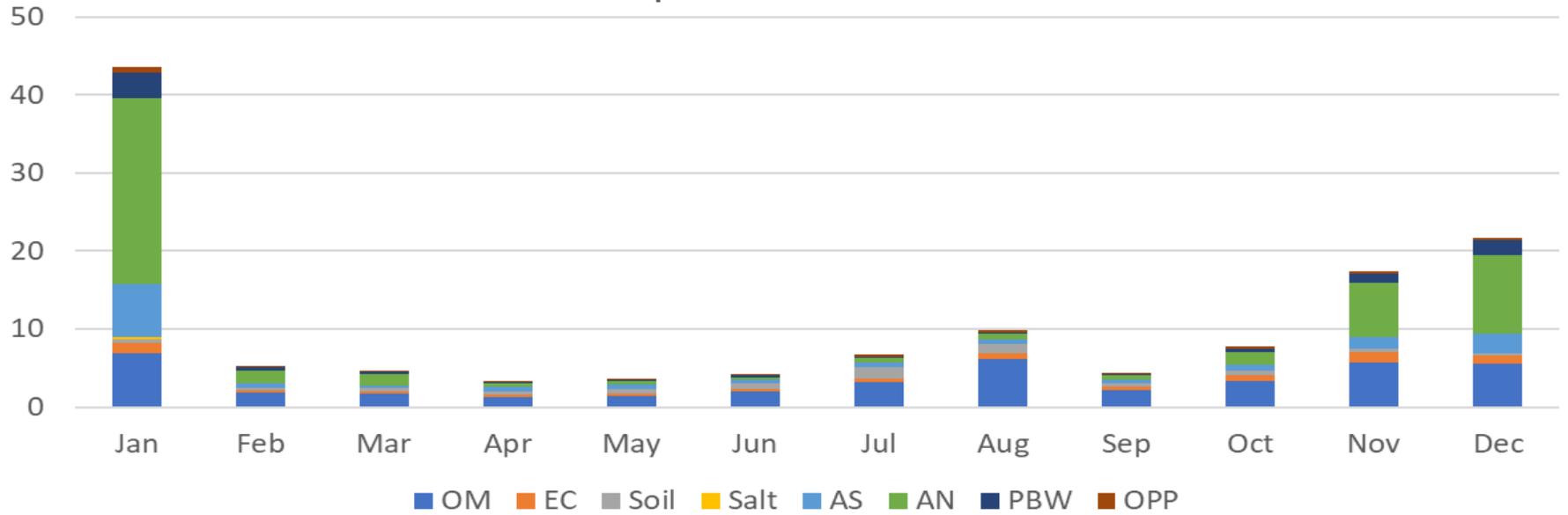
-4.60%



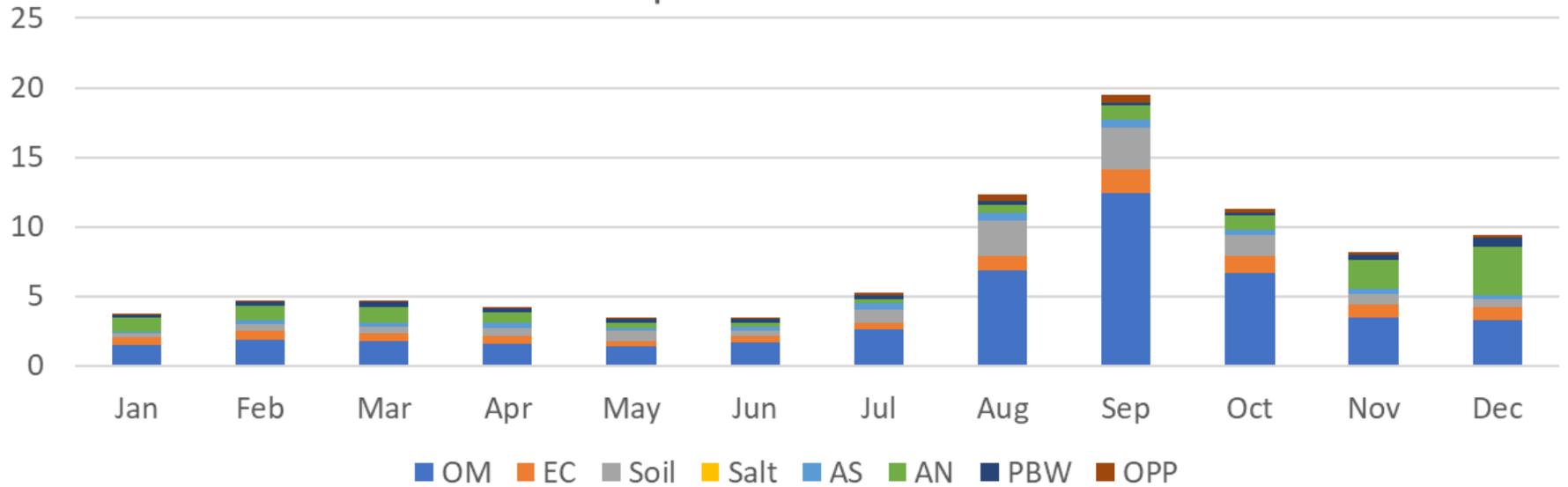
Speciation 2010-2020



## Speciation 2013



## Speciation 2020



# Summary

- Performed long-term speciation to understand the sources of PM<sub>2.5</sub> in the Treasure Valley, Idaho, USA
- Evaluated CSN-recommended speciation method
- Reduced bias
- Different seasonal patterns and dominant species:
  - ❖ Inversion: Highest PM<sub>2.5</sub> was observed in winter; nitrate
  - ❖ Wildfire & other fires: Highest PM<sub>2.5</sub> was observed in Summer & Fall; Organic Carbon
- Future work: Receptor modeling & chemical transport modeling