



APSORC17



Analysis of ‘High-Fired’ Plutonium Oxide in Tissues of Occupationally Exposed Workers

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*“Learning from Plutonium
and Uranium Workers”*





U.S. Transuranium and Uranium Registries

- Federal grant program funded by U.S. Department of Energy since 1968
- Today's mission is to follow up **occupationally exposed individuals**, from exposure through full lifespan, by studying the biokinetics (uptake, translocation, retention), and dosimetry of the actinides (Pu, Am, and U)



USTUR Registrants

- Voluntary tissue donors (posthumous):
 - whole-(44) and partial-body (305) donations
- Former nuclear workers from DOE sites
- Documented radiation exposure and work history
- Acceptance criteria:
 - i. actinide internal deposition of ≥ 74 Bq (2 nCi)
 - ii. external dose to whole body ≥ 0.1 Sv (10 rem)



USTUR: What Do We Do?





Tissue Sample Digestion

- HNO₃ – HCl – HF, 9:3:4 (ml) Soft tissues
- HNO₃ – HCl, 10:3 (ml) Bones
- Microwave unit CEM MARS 5
- Vessels EasyPrep[®]
- Temperature 220 °C
- Pressure 40 bar
- Time 30 min

Actinide Radiochemical Separation





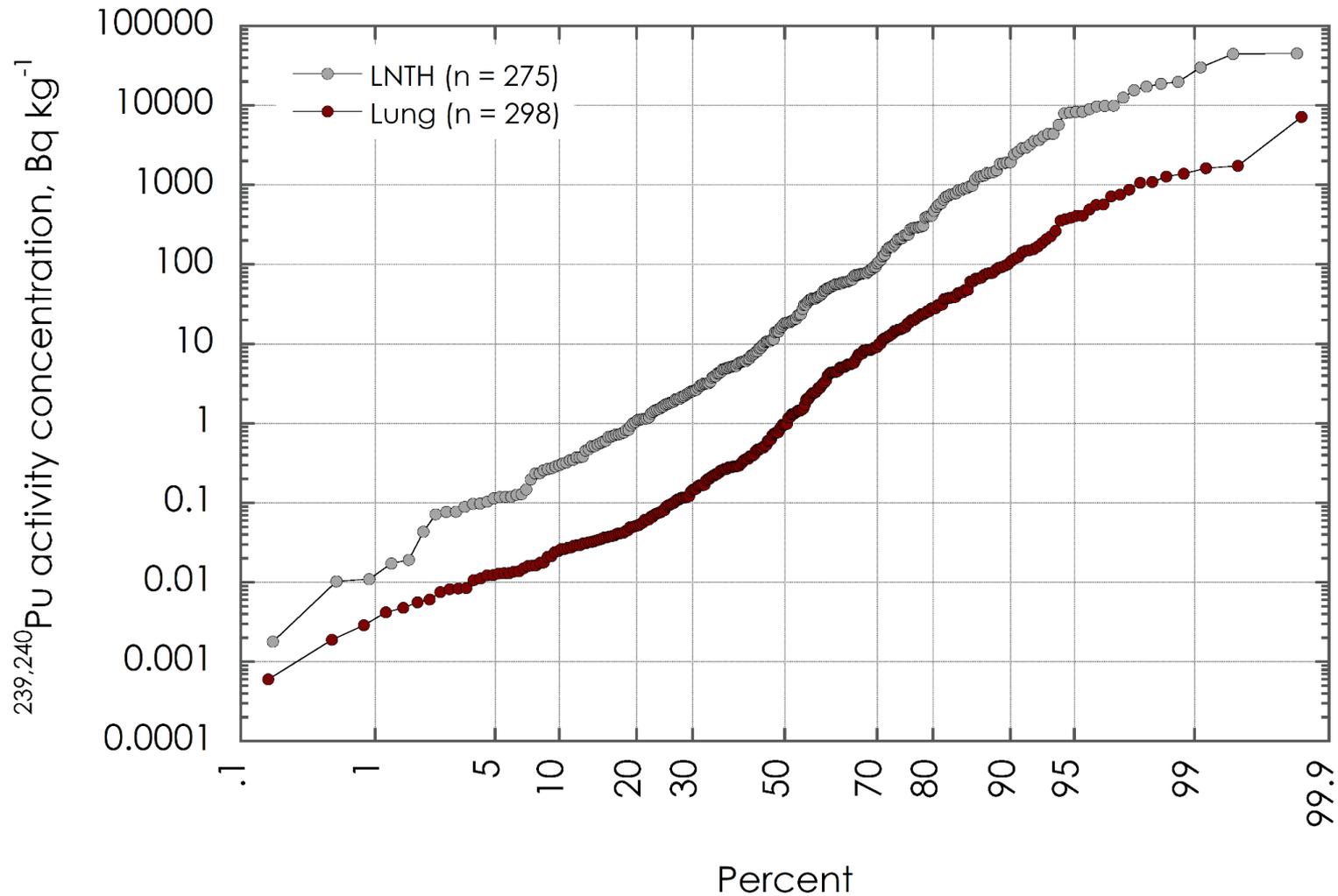
Actinide Electrodeposition

- Phoenix[®] EP-12 Series electrodeposition unit
- Na₂SO₄ electrolyte solution
- 1-h electrodeposition @ 0.75 A
- a-source: 5/8" stainless steel disk (planchet)



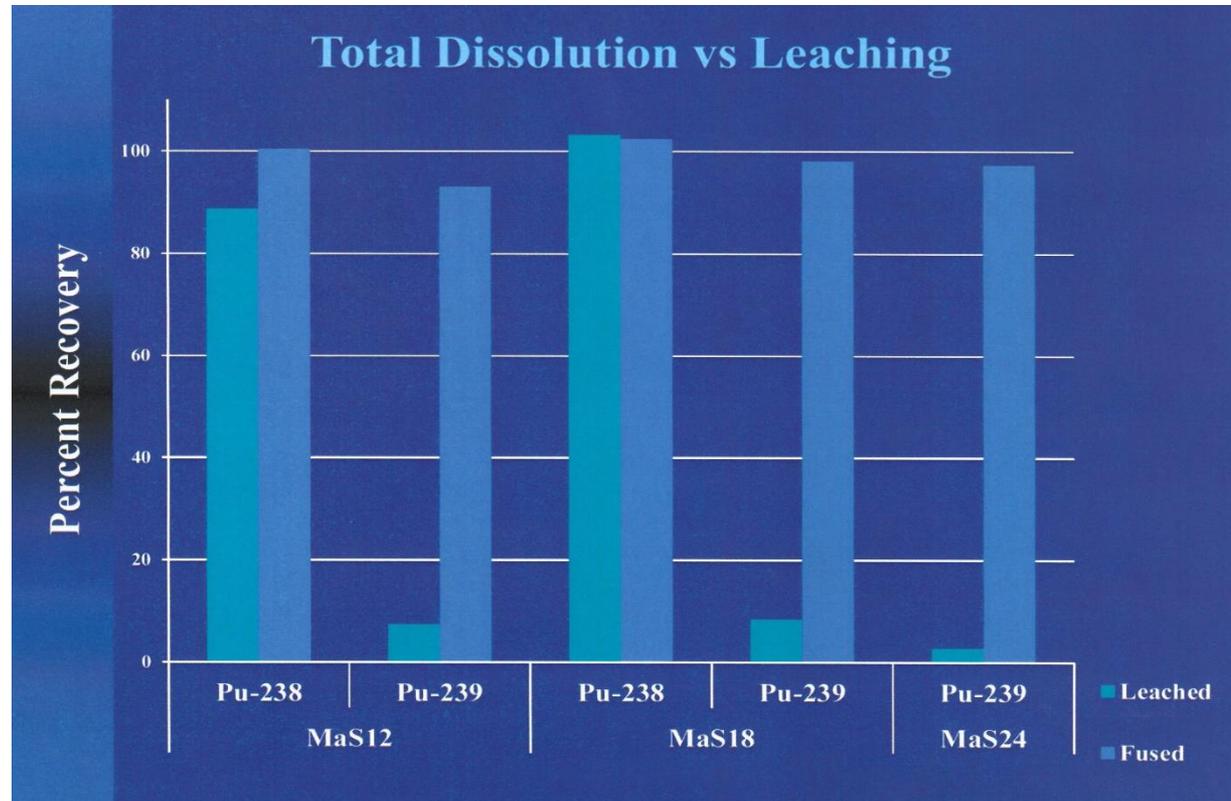


^{239}Pu in Respiratory Tract Tissues





Wake-up Moment



David S. Sill (Radiological and Environmental Sciences Laboratory, U.S. DOE) *In Search of... The Lost Art of Analytical Radiochemistry... The Last Crusade*. 60th Radiobioassay and Radiochemistry Measurements Conference (RRMC), Knoxville, TN, October, 2014

<http://www.rrmc.info/rrmc-60/rrmc-60-002p.pdf>



Method Validation at USTUR

Standard Reference Materials from NIST

- SRM 4351 (Human Lung)
- SRM 4352 (Human Liver)
- SRM 4356 (Human Bone)

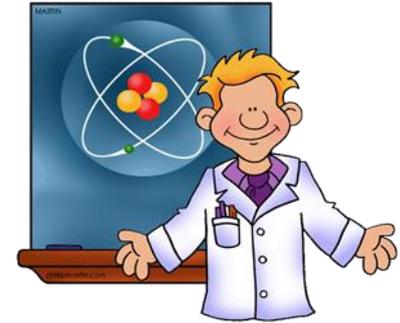
Split and spiked samples



No information on actinide material type!



Motivation



- Validation of USTUR tissue analysis protocol for 'high-fired' plutonium oxide ($^{239}\text{PuO}_2$)



Why 'High-Fired' Plutonium?

- Primary radionuclide of exposure: ^{239}Pu
- Primary material type: Type S (insoluble)
- Primary route of intake: Inhalation
- Deposition site: Respiratory tract



MAPEP-11-MaS24 Soil

- Prepared: Radiological and Environmental Sciences Laboratory (RESL)
- Reference values (1 pCi = 37 mBq):

^{241}Am	^{239}Pu	^{234}U	^{238}U
Concentration acceptance range, pCi g ⁻¹			
1.16 - 2.15	1.85 - 3.44	3.32 - 6.19	3.49 - 6.46
Mean \pm 95%CI, pCi g ⁻¹			
1.66 \pm 0.02	2.65 \pm 0.04	4.76 \pm 0.14	4.96 \pm 0.15

- Plutonium: 'high-fired' $^{239}\text{PuO}_2$ @ 900 °C

Results: Chemical Recovery of Tracer

Sample		Tracer recovery, %		
Size, g	N	²⁴³ Am	²⁴² Pu	²³² U
Digestion with HNO ₃ – HCl – HF				
0.5 × 2	4	33.4 ± 11.7	101.6 ± 5.9	96.7 ± 6.2
1.0	8	30.8 ± 3.3	99.9 ± 2.7	97.4 ± 5.6
2.0	4	42.3 ± 5.6	101.8 ± 5.7	101.6 ± 6.7
Mean ± 95%CI	16	34.3 ± 3.6	100.8 ± 1.8	98.3 ± 3.0
Digestion with HNO ₃ – HCl				
1.0	6	23.0 ± 4.2	101.6 ± 8.8	91.0 ± 7.5
Residue [†]	6	98.4 ± 1.5	101.7 ± 4.2	93.9 ± 7.4

† - additionally treated with HF

Soil matrix effect on Am separation !

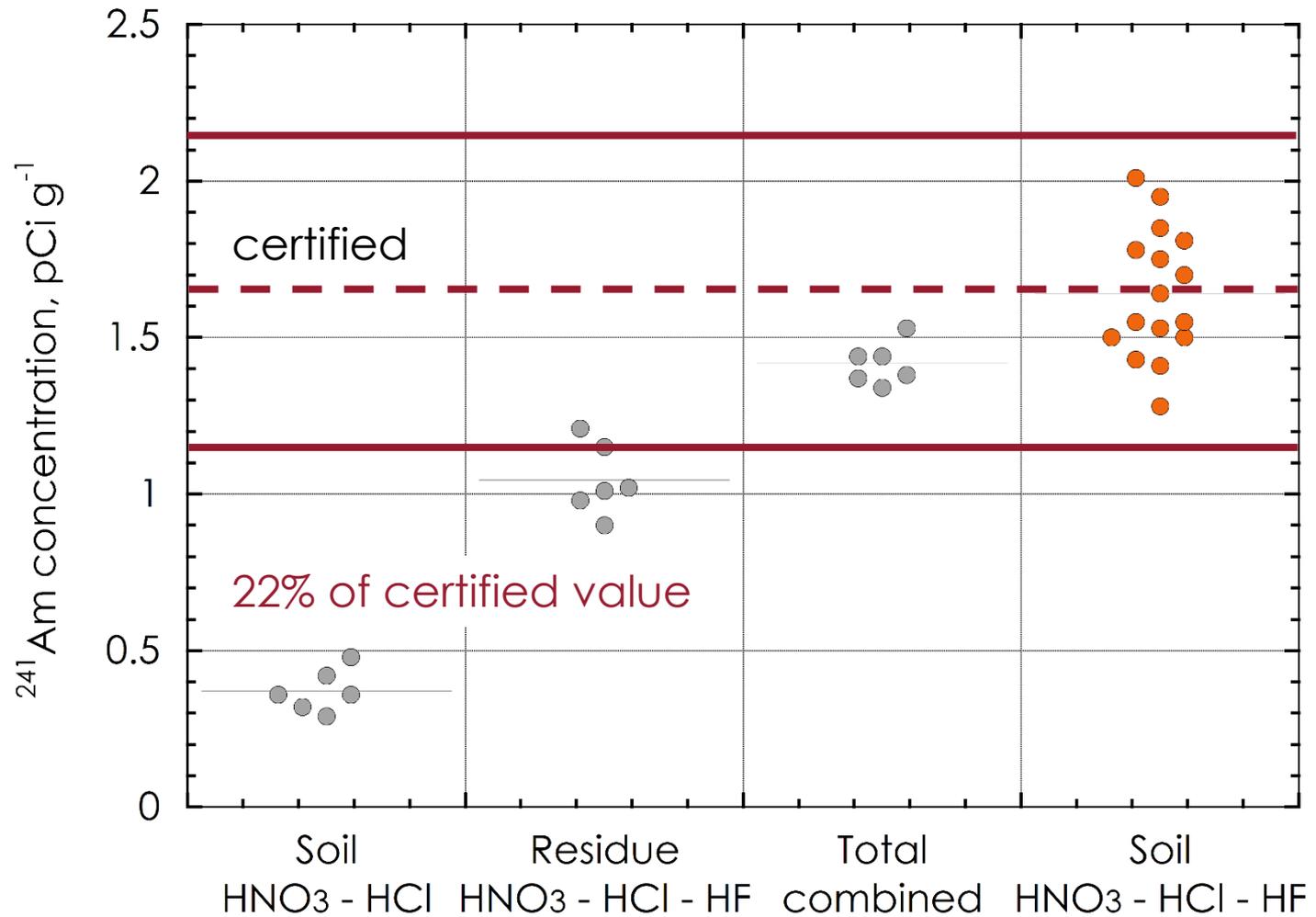
Results: Analyte Activity Concentration

Sample		Analyte concentration, pCi g ⁻¹		
Size, g	N	²⁴¹ Am	²³⁹ Pu	²³⁸ U
Digestion with HNO ₃ – HCl – HF				
0.5 × 2	4	1.58 ± 0.16	2.38 ± 0.17	4.62 ± 0.45
1.0	8	1.73 ± 0.18	2.66 ± 0.15	4.68 ± 0.40
2.0	4	1.50 ± 0.32	2.48 ± 0.14	4.47 ± 0.27
Mean ± 95%CI	16	1.64 ± 0.11	2.54 ± 0.10	4.61 ± 0.20
Digestion with HNO ₃ – HCl				
1.0	6	0.37 ± 0.07	0.14 ± 0.01	1.39 ± 0.19
Residue [†]	6	1.05 ± 0.12	2.42 ± 0.10	3.28 ± 0.14
Total combined	6	1.42 ± 0.08	2.56 ± 0.06	4.68 ± 0.14

† - additionally treated with HF

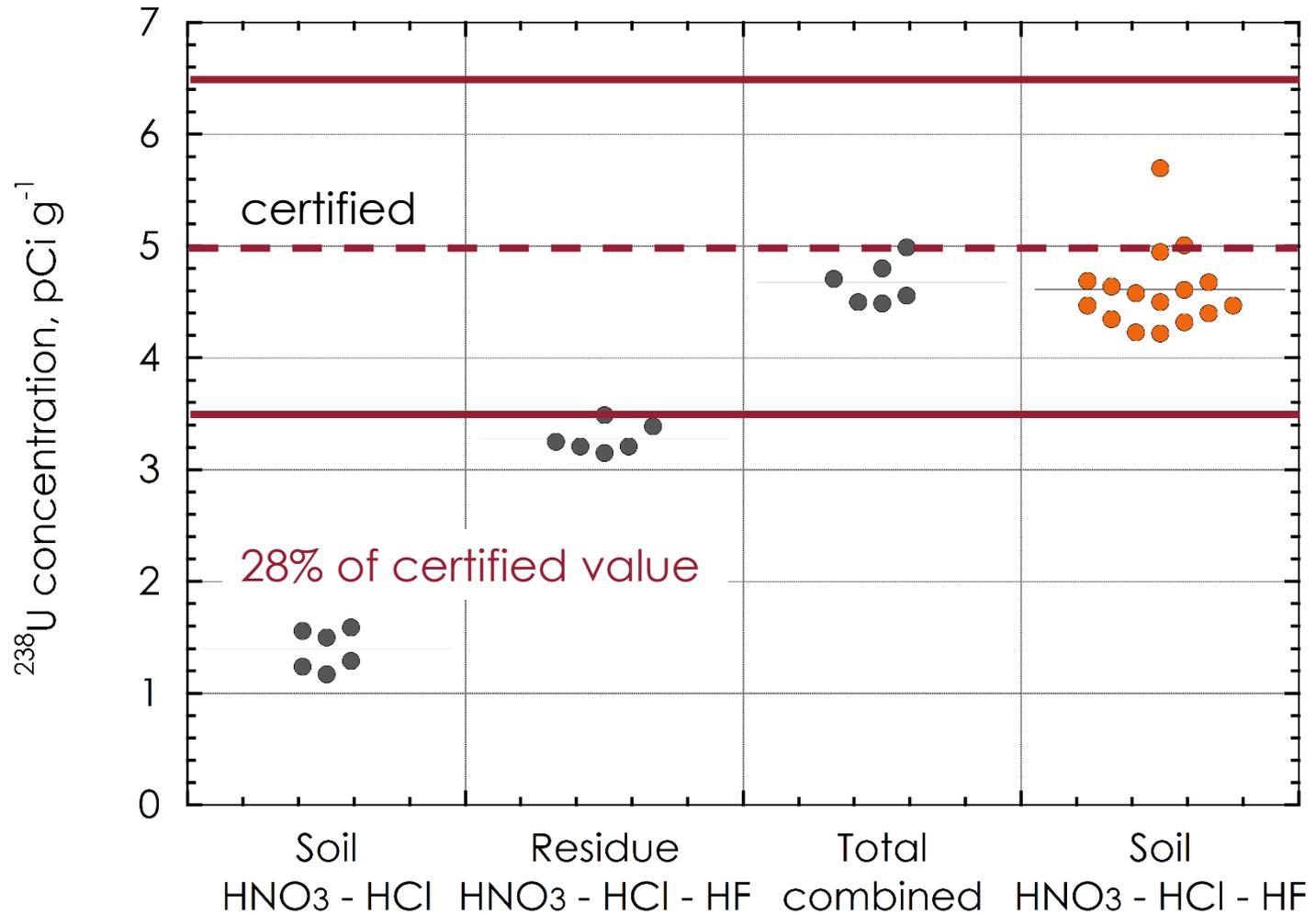


^{241}Am : Measured vs Certified





^{238}U : Measured vs Certified





Conclusions

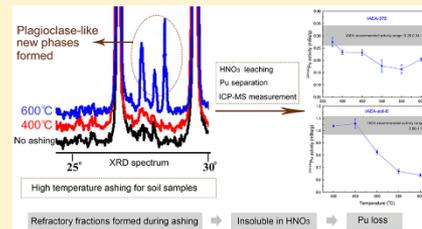
- USTUR tissue analysis protocol was **successfully** validated for the analysis of 'high-fired' PuO₂



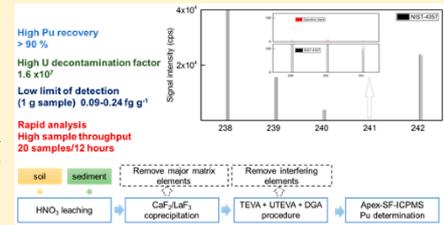
- Microwave-assisted acid digestion with HNO₃ – HCl **does not** allow for total plutonium dissolution
- Hydrofluoric acid (HF) is **critical** for analysis of insoluble plutonium material

Mystery!

ABSTRACT: An acidic leaching method using HNO_3 is widely employed to release the global fallout Pu from soil samples for further chemical separations in radioecology and toxicology studies and in many applications using Pu as a useful tracer. In the method's sample ash treatment step to decompose organic matter in soil, various ashing temperatures (400–900 °C) are used; however, the effect of ashing temperature on the accurate Pu analysis has not been well investigated. In this study, two standard reference soils (IAEA-soil-6 and IAEA-375) were used to determine the ashing temperature effect (from 375 to 600 °C) on the HNO_3 leaching method. The Pu analytical results of both standard reference materials showed that lower $^{239+240}\text{Pu}$ activity was observed when the ashing temperature exceeded 450 °C, and the $^{239+240}\text{Pu}$ activity continued to decrease as the ashing temperature was raised. Approximately 40% of the Pu content could not be leached out by concentrated HNO_3 after ashing for 4 h at 600 °C. The Pu loss was attributed to the formation of refractory materials, which are insoluble in HNO_3 solution. This hypothesis was confirmed by the XRD analysis of soil samples, which revealed that plagioclase-like silicate materials were formed after high-temperature ashing. To ensure Pu release efficiency in HNO_3 leaching, we recommend 450 °C as the ideal ashing temperature. This recommendation is also useful for analysis of other important artificial radionuclides (e.g., ^{137}Cs , ^{90}Sr , ^{241}Am) for which an ashing process is needed to decompose the organic content in soil samples.



ABSTRACT: Plutonium is extensively studied in radioecology (e.g., soil to plant transfer and radiological assessment) and geochemistry (e.g., sediment dating). Here, we reported a new chemical separation method for rapid determination of Pu in soil and sediment samples, based on the following investigations: extraction behaviors of interfering elements (IEs, for inductively coupled plasma mass spectrometry (ICPMS) measurement) on TEVA resin; decontamination of U using TEVA, UTEVA, and DGA resins; and the impact of coprecipitation on Pu determination. The developed method consists of four steps: HNO_3 leaching for Pu release; $\text{CaF}_2/\text{LaF}_3$ coprecipitation for the removal of major metals and U; the proposed TEVA + UTEVA + DGA procedure for the removal of U, Pb, Bi, Tl, Hg, Hf, Pt, and Dy; and ICPMS measurement. The accuracy of this method in determining $^{239+240}\text{Pu}$ activity and $^{210}\text{Pu}/^{239}\text{Pu}$ and $^{241}\text{Pu}/^{239}\text{Pu}$ isotopic ratios was validated by analyzing five standard reference materials (soil, fresh water sediment, and ocean sediment). This method is characterized by its stable and high Pu recovery (90–97% for soil; 92–98% for sediment) and high decontamination factor of U (1.6×10^7), which is the highest reported for soil and sediment samples. In addition, the short analytical time of 12 h and the method detection limits, which are the lowest yet reported in literature, of $0.56 \mu\text{Bq g}^{-1}$ (0.24 fg g^{-1}) for ^{239}Pu , $1.2 \mu\text{Bq g}^{-1}$ (0.14 fg g^{-1}) for ^{240}Pu , and 0.34 mBq g^{-1} (0.09 fg g^{-1}) for ^{241}Pu (calculated on the basis of a 1 g soil sample) allow the rapid determination of ultratrace level Pu in soil and sediment samples.



“Approximately 40% of the Pu content could not be leached out by concentrated HNO_3 after ashing for 4 h at 600 °C. The Pu loss was attributed to the formation of refractory materials, which are insoluble in HNO_3 solution”

“The developed method consists of four steps: HNO_3 leaching for Pu release; $\text{CaF}_2/\text{LaF}_3$ coprecipitation ...; the proposed TEVA + UTEVA + DGA procedure; and ICPMS measurement. ...This method is characterized by its stable and high Pu recovery (90–97% for soil; 92–98% for sediment)”



Questions?

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