

Refrigerant Selection

If you have not investigated the impact of different refrigerants on the properties of the last 2 cycles you've analyzed, I suggest you do. It makes a big difference!

As we'll find out, refrigerants are everywhere. (Shower can, cooking spray)

Some of you may remember the switch from R12 $\rightarrow$ R134a in automotive AC systems. Before we discuss this, let's talk about the history of refrigerants

$\rightarrow$ Pre 20th century the state of the art or "modern" refrigeration used ice.

$\rightarrow$ The primary unit of refrigeration is the ton: $1 \text{ ton} = 3517 \text{ W}$.
Which is the amount of cooling provided by melting 1 ton of ice in 1 hour.

$\rightarrow$ Think about this for a minute... a building like this could easily have 100 tons of AC cooling capacity: $100 \text{ tons} \frac{\text{ice}}{\text{hour}}$

$$\rho_{\text{ice}} = 916.7 \frac{\text{kg}}{\text{m}^3}, m_{\text{ice}} = 91,718 \text{ kg}, V_{\text{ice}} = 98.96 \text{ m}^3$$

Football field is $100 \text{ yd} \times 53.5 \text{ yd} = 1''$ thick layer of ice melted/hour
Basketball court $94 \text{ ft} \times 50 \text{ ft} \approx 9''$ thick layer of ice melted/hour

$\rightarrow$ The invention of mechanical refrigeration with compressors changed everything! But what to use as refrigerant?

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Good refrigerant characteristics depend on the demands of a specific application:

- ✓ 1) Positive evaporator gauge pressure $\Rightarrow$ prevents air from leaking into system
- ✓ 2) Moderate condenser pressure $\Rightarrow P_{\text{sat}}$ increases almost exponentially with T_{sat}
- ✓ 3) Triple point & critical point temperatures m^3/kg
- ✓ 4) Low specific volume ^(high density) @ compressor inlet: $\text{Disp} = \frac{\dot{m}}{\rho_{\text{inlet}}} \Rightarrow \frac{\dot{V}_{\text{in}}}{\rho_{\text{inlet}}}$
- ✓ 5) high latent heat of vaporization: $h_g = h_f + h_{fg}$
- ✓ 6) high dielectric strength (ability to resist flow of electricity) for cooling electronics
- ✓ 7) Compatibility with lubricants
- ✓ 8) Non-toxic
- ✓ 9) Non-flammable
- ✓ 10) Inertness/stability $\Rightarrow$ prevents degradation of components
- ✓ 11) Ozone depletion / Global Warming Potential

$\rightarrow$ A huge list & it's difficult to find or make a fluid that addresses all of those

$\rightarrow$ Ammonia was used from the late 1800's $\rightarrow$ 1920s in households, still used in industry

$\rightarrow$ 1928 Thomas Midgley Jr. & Charles Frank Kettering invented Freon ^{Problem: toxic} R12 @ GM.

$\rightarrow$ also invented leaded gasoline which doubled amount of gas we can make from a barrel of oil $\rightarrow$ but a lot of the lead is ^{lost}

Invented electronic ignition systems

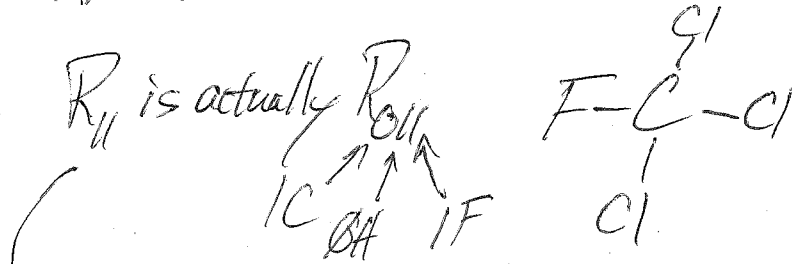
$\rightarrow$ Chlorofluorocarbons

$\rightarrow$ This division was later bought out by DuPont. R11 & R12 used until 1989.

We still use the DuPont naming scheme for these refrigerants WBZ
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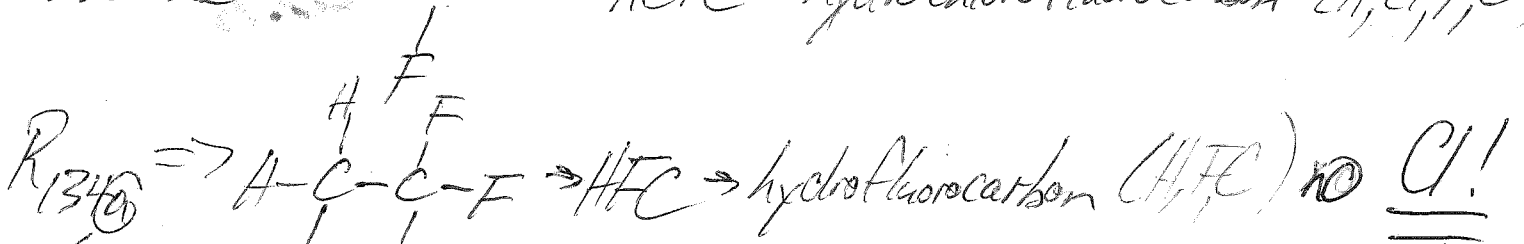
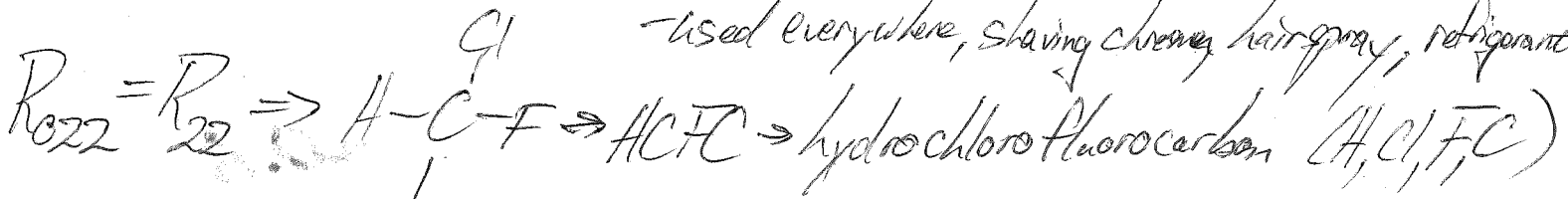
R_{xyz}
 x = # of carbon atoms - 1
 y = # of fluorine atoms
 z = # of hydrogen atoms

All of these have Chlorine atoms



$\rightarrow \text{CFC} \rightarrow$ chlorofluorocarbon (C, F, C)

- used everywhere, shaving cream, hair spray, refrigerant



Other ways to make same combo $\begin{array}{c} \text{F} \quad \text{F} \\ | \quad | \\ \text{H}-\text{C}-\text{C}-\text{H} \\ | \quad | \\ \text{F} \quad \text{F} \end{array} \neq$ very different, ^{separation required} we'll talk about this some in statistical mechanics

- Why is Cl bad? $\rightarrow$ Destroys Ozone $\text{O}_3 \leftarrow$ protects us from UV radiation

- By making R-11, R-22 etc very stable, it survives up into the highest regions of the atmosphere where cosmic rays break it up, causing reactions $\text{Cl} + \text{O}_3$

- CFCs outlawed in 1989

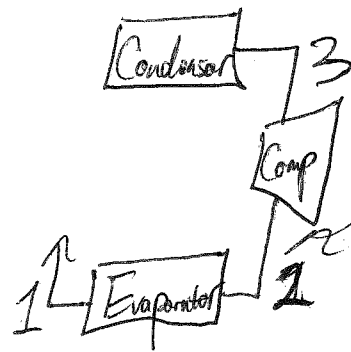
- HCFCs will probably be outlawed in next 10 years $\leftarrow$ huge costs to changeover equipment

- HFCs have 0 Ozone Depletion ^{Potential (ODP)} but some overall GWP $\leftarrow$ Global Warming potential

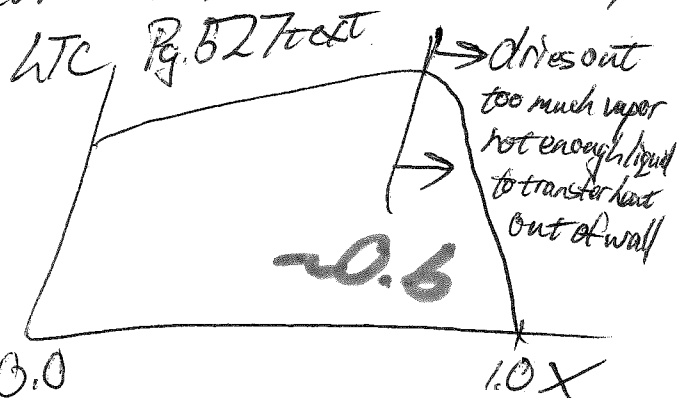
- See Figure ^{9.1} in text pgs 578 $\rightarrow$ 579

- Conclusion: we're using better chemicals with lower ODP & GWP, but we still have to make these systems efficient to reduce CO_2 & its GWP

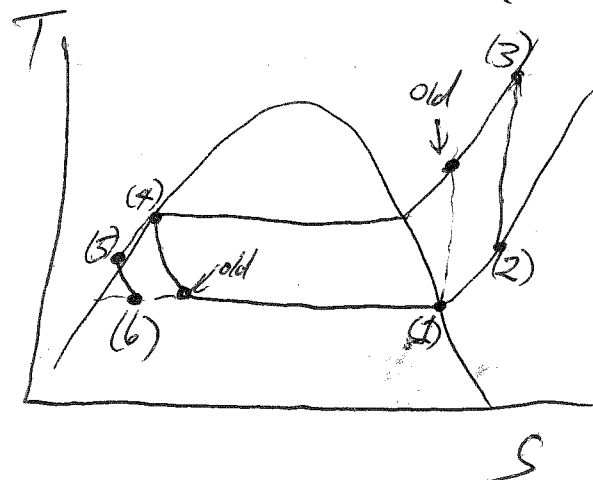
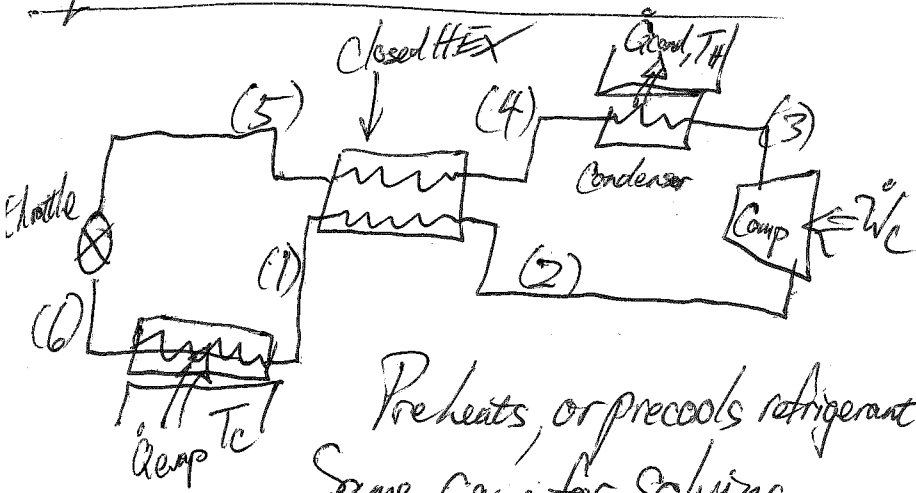
ME301 Sp 2011
 18.4
 32.3
Vapor Compression Cycle Modifications Assign 9.A → 13
 or superheated vapor
 We want saturated vapor @ evaporator exit.
 $x \rightarrow 1.0$



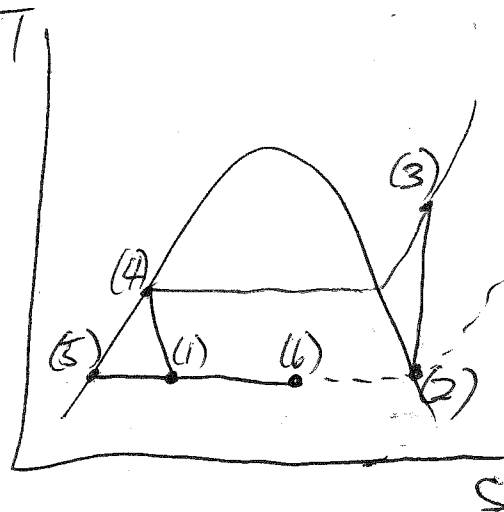
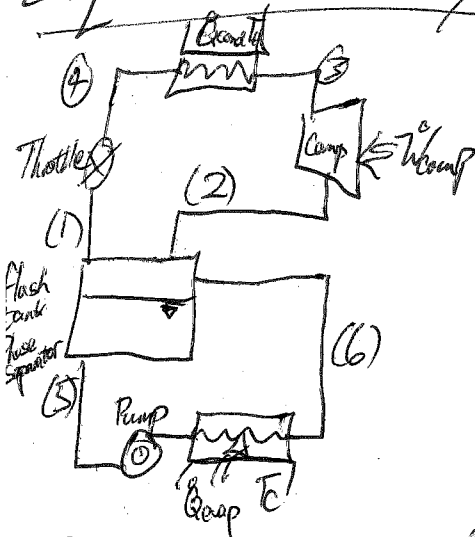
But, heat transfer effects that we have been ignoring in this class play a factor ← will be covered in ME564



Liquid Suction A/E X Modification



Liquid Over-Feed System Modification



Multiple compressor cycles & combinations of these modifications. More complexity, still the same game of defining all of the states, then analyzing.

Can change pump speed to increase ϵ_{evap} , could also connect several evaps to this.